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Two Part Report on Dr. Dittschmidt's Oil Circulation Process

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T-450

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I. DISCUSSION OF CARBON MONOXIDE-HYDROGEN  
SYNTHESIS IN BERLIN, JULY 1, 1941

Dr. Lindt has discovered the fused iron catalyst in 1937 to 1938.

At that time deposition of carbon caused difficulties when operating under high pressures in the vapor phase.

Experiments in the liquid phase were begun in 1934 and resulted in the development of the oil circulation processes, in which oil and gas were led concurrently over a fused iron catalyst. It is important here that operations be conducted in the boiling phase, produced by the return of the relatively low boiling oil-constituents. The work was originally conducted at 100 atm, and subsequently at middle pressures in two stages.

With a 240°C reactor inlet temperature and 230°C reactor outlet, a total of 150 grams of primary products (liquid and gasol) were obtained in the two stages per mole of the ideal gas, with a 50% conversion in each stage. The composition of the primary products was:

40% primary gasoline, c.n. 62-68  
c.n. with 0.1% lead was 85  
2% oxygen content, removable by washing with water.

20% middle oil	cetane No. 78
20% hard paraffin	m.p. 95°C
15% gasol (no C <sub>7</sub> )	with 85% olefines
5% alcohols	with 25% methanol, 50% ethanol, 25% higher alcohols, acetaldehyde, acetone, etc.

The space-time yield was

$30 \text{ g/lit cat/hr} = 0.76 \text{ lit cat/day}$

Experiments at 25, 300, 150 and 200 atm. have shown that the oxygen content rises with increasing pressure, and at 150 atm. very much low boiling alcohols and very little higher alcohols are obtained.

The process can be used for alcohol production when operating with a low conversion and with a hydrogen-rich synthesis gas.

With a CO:H<sub>2</sub> ratio of 1:1, at 150 atm. and 200-220°C, with a 20-30% conversion, 48.5% alcohols was found in the total liquid formed which consisted of

- 3.5% methanol
- 21.0% ethanol
- 13.0% propanol
- 5.5% C<sub>4</sub>-C<sub>11</sub> alcohols
- 2.5% C<sub>12</sub>-C<sub>20</sub> alcohols

The 32.5% hydrocarbons consisted of

- 26.5% gasoline
- 3.5% middle oil
- 3.5% over 500°C.

The 12% fatty acids consisted of:

- 11% water solubles
- 5% C<sub>8</sub>-C<sub>11</sub>
- 2% C<sub>12</sub>-C<sub>20</sub>.

The alcohols disappear in the liquid fraction

- (a) when the conversion is high,
- (b) when carbon is deposited upon the catalyst.

IBM Reel 13, Bag 3343, Item 2, p. 150

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VI. DISCUSSION OF JULY, 1942

The process is practically identical with Michael's four process with the exception that the catalyst is arranged in a fixed bed making the separation of the catalyst unnecessary.

Maximum reaction capacity	At present up to 1.5 atm
Space time yield	0.72 kg/hr/lit catalyst
Conversion	87% in two stages
Oil circulation considerably smaller than in Michael's	CO <sub>2</sub> formation quite predominate over the formation of water
Composition of products	150 gram total products of which 40% gasoline 30% gas oil 50% hard paraffin wax 15% gasol 15% alcohol

(It is noted that the water formed amounts to 10% of the above and the total amount of liquid formed adds then up to 110%.)

ANALYSIS OF HYDROCARBONS

C No.	Per Cent by Wt. in 150 grams	Per Cent Olefines <sup>a</sup>	
	C <sub>6</sub>	8.6%	75-80%
	C <sub>7</sub>	7.4	75-80
50-50°G	C <sub>8</sub>	6.6	60-75
50-75	C <sub>9</sub>	6.6	60-70
75-100	C <sub>10</sub>	6.5	55-60
100-125	C <sub>11</sub>	6.2	50-55
125-150	C <sub>12</sub>	5.0	50-55
150-175	C <sub>13</sub>	3.7	40-50
175-200	C <sub>14</sub>	3.2	35-40
200-225		3.1	35-40
225-250		3.6	
250-275		1.9	
275-300		3.3	
Hard Paraffin Wax		11.4	
Alcohols		20.0	
		5.7	

<sup>a</sup>The olefines: 50% branched, 50% straight chain